Density D= m/v

Density

- Density is the amount of matter in a given amount of space
- Density is how closely packed particles of matter are.

Density

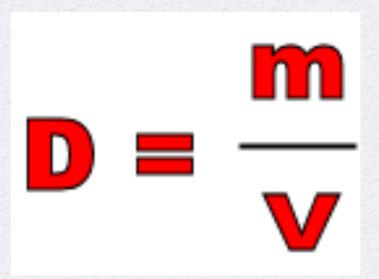
- How do we find the mass of an object?
 - Scale (Triple beam balance)
- How do we find the volume of an object?
 - How do we find the volume of a liquid or a gas?
 - If it is a rectangular solid
 - Length · width · height



- If it is irregularly shaped use displacement
 - Use a graduated cylinder and see how high the water moves up when the object is put in

Formula for Density and its units

Use g/mL or g/cm³





You can tell by observation which object is more dense

- Water has a density of 1 g/mL. If an object floats, is it more or less dense than water?
- When an object heats up, does it become more or less dense?

Examples

- Submarines
- http://science.howstuffworks.com/ submarine1.htm

What evidence do we have that particles of matter are in motion?

- Air freshener spreads out to every part of a room.
- Vanilla escapes from a balloon.
- Food coloring spreads out in water.

Temperature affects how fast particles of matter are moving

- When particles are heated up they will speed up, spread apart and become less dense.
- That's why warm air rises, and warm water rises.

Examples

http://www.miamisci.org/af/sln/phases/ index.html

Substances expand when they heat up and contract when they cool down.

- Cracks in the sidewalk are made so that they can expand in the hot summer sun and not crack.
- The liquid in a thermometer expands and moves up when it heats up.

Solid Liquid Gas